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| 11. | I/We request the grant of a patent on the basis of this application. | | |
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ELECTROCHEMICAL DETECTOR FOR METABOLITES IN PHYSIOLOGICAL FLUIDS

Introduction

The concentrations of metabolites in body fluids, particularly in blood, are key indicators to the physiological state of the body. Monitoring of these metabolites is therefore desirable when disease is present or suspected. For example, the level of glucose in blood provides information on the status of a diabetic patient.

There exist many analytical techniques for measuring metabolites in physiological fluids. These fall into three main categories: (i) procedures performed in a specialised laboratory, (ii) techniques executed by a healthcare professional at the bedside or (iii) devices designed for personal use. Glucose testing is a good example which falls into all three areas. Frequent testing of blood glucose is critical to controlling diabetes. The Diabetes Control and Complication Trial (DCCT), a 10-year study carried out by the National Institutes of Health, found that people who test their blood glucose four times or more each day can lower their risk of developing eye, kidney and nerve diseases and high cholesterol. Medical doctors recommend glucose measurements should be made a number of times per day until stabilized.

Background to the Invention

The invention relates to the measurement of metabolites in body fluids using a novel electrochemical sensor system. In one embodiment, the sensor may be used for measuring glucose and other metabolites in interstitial fluid (fluid which exists between cells). When used in combination with a suitable method of extracting interstitial fluid from skin, the sensor offers a highly effective and essentially non-invasive means of monitoring glucose and other metabolites.

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Direct electrochemical detection of glucose in blood is notoriously difficult, due the many large molecules that foul the electrode. By performing the analysis in interstitial fluid instead, this problem is mitigated as most of the plasma protein molecules are too large to pass through the capillary walls into the interstitial area. Furthermore, the invention incorporates electrochemical cleaning of the electrode surface to remove fouling agents, and produces multivariate responses which allows the effects of any remaining interferents to be negated using subsequent data processing.

Technical Description

Sensor Construction

The device consists of a noble metal electrode such as platinum or gold in combination with a reference electrode such as silver/silver chloride. A third counter may also be used alongside the other electrodes. The working electrode is surrounded by a highly alkaline microenvironment, to aid electrode cleaning. This environment can be achieved using an alkaline polymer electrolyte or a hydrogel impregnated with basic electrolytes (an example is agarose gel with NaOH). Other materials may also be used for this purpose. Alternatively, volumes of basic electrolytes such as NaOH can be used to dilute the fluid prior to electrochemical analysis.

Electrochemical Measurements

The sensor assembly is placed into the sample matrix of interest, for example interstitial fluid containing glucose. The three electrodes are electrically connected to a potentiostat instrument capable of performing voltammetry.

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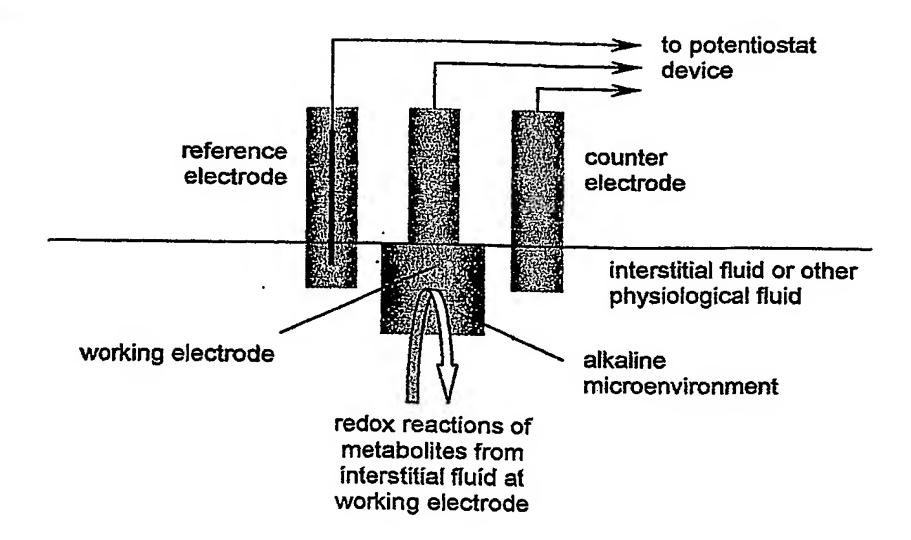


Figure 1: Diagram showing the experimental set-up of the sensor.

A potential is applied at the working electrode, following the waveform shown in Figure 2. This consists of two cleaning pulses, which clear the electrode of any electrochemical breakdown products from previous measurements, followed by a voltammetric sweep during which current measurement takes place. The whole measurement, which takes just a few seconds, results in a current versus potential response such as that shown in Figure 3.

Simple parameters extracted from the observed data, such as peak height, can be calibrated to give the concentration of individual metabolites of interest. However, further chemometric processing using, for example, the multivariate regression abilities of feed forward neural networks (Figure 4) can be used to provide more accurate measurements. Such chemometric methods also allow the quantification of more than one analyte from a single measurement, as each analyte is typically active a particular point in the voltammetric sweep.

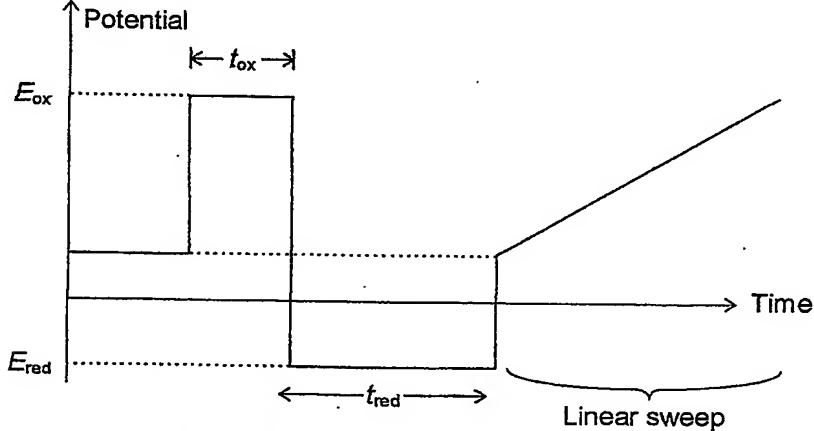


Figure 2: Applied potential waveform.

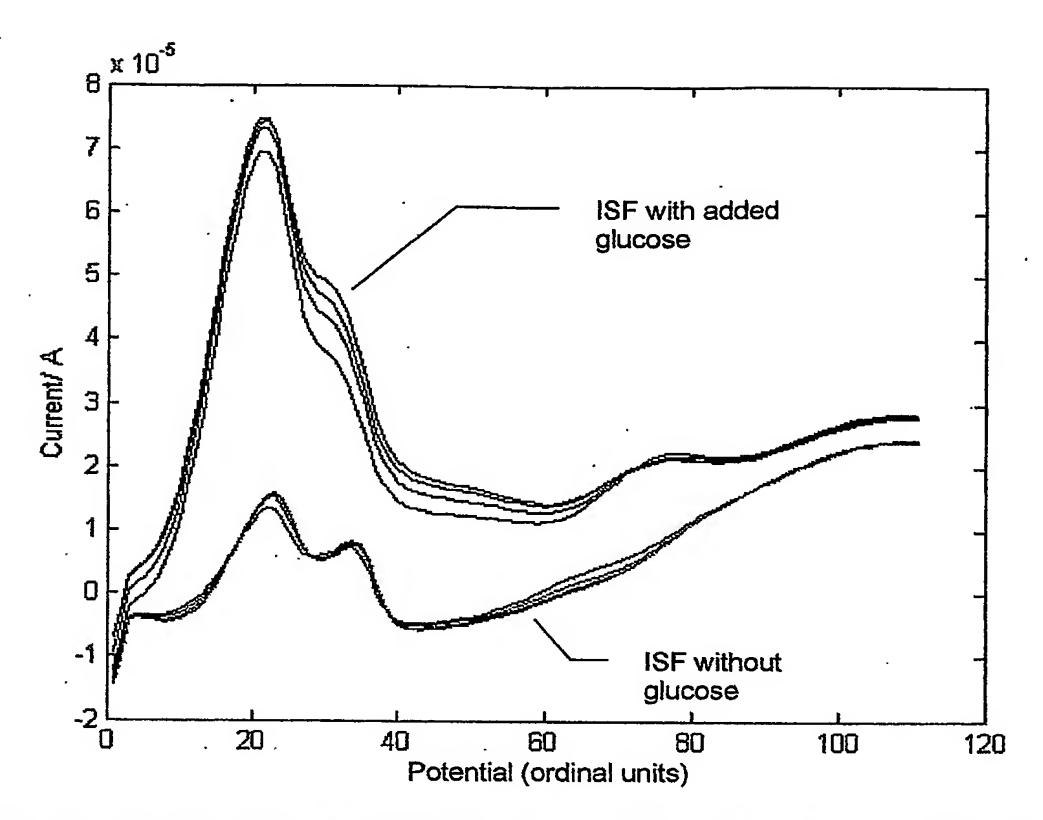


Figure 3: Example of current response acquired from the sensor, in this case for interstitial fluid spiked with glucose. The potential range scanned in this case was approximately -0.9V to 0.2V.

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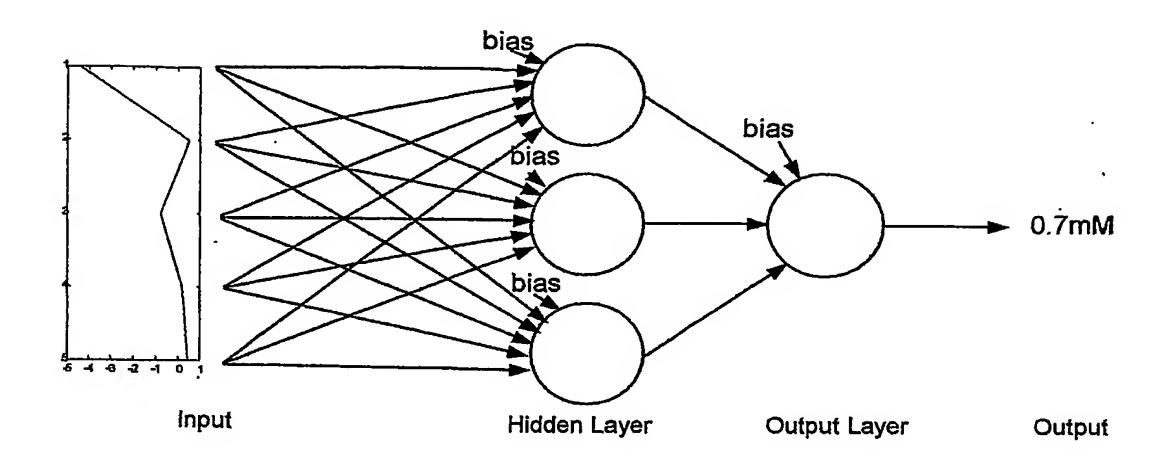


Figure 4: Neural network calibration of sensor data. In this case, the number of inputs to the network is optimized by reducing the number of points in the acquired voltammogram using linear algebra.

Benefits of Invention

The sensor device is capable of simultaneously measuring several analytes in body fluids in a single measurement cycle. Particularly useful features of the invention are:

- 1. High sensitivity for certain metabolites such as glucose, permitting application in small volumes, such as those associated with interstitial fluids.
- 2. The ability to self-clean between each measurement.
- 3. High stability of the sensor.
- 4. Opportunity to measure more than one metabolite simultaneously.
- 5. Speed of measurement (<5 seconds).
- 6. Potential small size of low cost of equipment, allowing embodiment of the invention as a patient operated device or autonomous system.